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Synthesis and Fluorescence Properties of 2,6-Diaryl-4-(2-substituted thienyl-5-yl)pyridines

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Synthesis and Fluorescence Properties of 2,6-Diaryl-4-(2-substituted thienyl-5-yl)pyridines

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INTRODUCTION

The structure of fluorescent compounds is important in designing luminescent materials for organic light-emitting diodes and dye lasers, fluorescent labeling reagents for HPLC or CE, and so forth. In our earlier work, 2,5-di(4-pyridyl)thiophene and 4-(5-substituted 2,2'-bithienyl-5'-yl)pyridines were shown to be useful materials as a nitrogen laser dye and labeling reagents, respectively, because of their strong fluorescence and photostability.

RESULTS

2,6-Diphenyl- and 2,6-di(2-thienyl)-4-(2-substituted thienyl-5-yl)pyridines (1) were prepared from the corresponding 1,5-diaryl-3-(2-substituted thienyl-5-yl)-1,5-pentadiones (2).

$$R_1$$
—COCH₃ + R_2 —CHO
$$R_1$$

$$R_1$$

$$R_2$$

$$R_1$$

$$R_1$$

$$R_1$$

$$R_1$$

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TABLE I

| | | | Yield (%) | |
|--------------|-----------|-------------------------|-----------|----|
| | R_1 | $ m R_2$ | 1 | 2 |
| a | phenyl | 2-thienyl | 53 | 72 |
| b | phenyl | 2,2'-bithienyl- 5 -yl | 13 | 25 |
| \mathbf{c} | 2-thienyl | 2-thienyl | 22 | 50 |
| d | 2-thienyl | 2,2'-bithienyl- 5 -yl | 11 | 24 |

¹H and ¹³C NMR, HRMS-FAB spectra, and elemental analysis of compounds **1a-d** and **2a-d** were found in good agreement with their structures. As shown in Table II, the wavelengths of maximum absorption of **1b-d** in 60% MeOH were similar to those of the corresponding 2,6-unsubstituted 4-(2-thienyl)pyridine (**1e**) and 4-(2,2'-bithienyl-5-yl)pyridine (**1f**); Both molar absorptivities and fluorescence intensities of **1a-d** were larger than those of the corresponding **1e** and **1f**, respectively. On the other hand, in strong acidic 60% MeOH, no shift in the absorption maximum based on the protonated form for **1c** and **1d** was observed.

TABLE II Absorption and Fluorescence Properties of 1a–1f in 60% MeOH

| | $\lambda_{abs} (nm)$ | ε | pKa | $\lambda_{\rm ex}~({\rm nm})^a$ | $\lambda_{\rm em}~({\rm nm})^a$ | $\mathrm{RFI}^{a,b}$ | τ (ns) |
|----|-----------------------|-------|------|---------------------------------|---------------------------------|----------------------|--------------------------------|
| 1a | 263 | 35700 | 2.87 | 310 | 370 | 285 | $0.87 \pm 0.08, 3.21 \pm 0.01$ |
| 1b | 353 | 31000 | 3 | 362 | 443 | 624 | 1.31 ± 0.01 |
| 1c | 298 | 40600 | | 306 | 402 | 193 | 1.55 ± 0.01 |
| 1d | 357 | 37000 | | 367 | 449 | 838 | 1.37 ± 0.01 |
| 1e | 295 | 16000 | 4.48 | 304 | 359 | 101 | $0.80 \pm 0.05, 2.00 \pm 0.48$ |
| 1f | 354 | 27400 | 4.44 | 359 | 438 | 17 | 1.14 ± 0.01 |

^aObtained from uncorrected spectra.

^bRelative fluorescence intensity.